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American Chemical Society and is provided to assist you in searching databases on STN. Any dissemination, distribution, copying, or storing of this information, without the prior written consent of CAS, is strictly prohibited. FILE COVERS 1907 - 5 Jun 2002 VOL 136 ISS 23 FILE LAST UPDATED: 3 Jun 2002 (20020603/ED) This file contains CAS Registry Numbers for easy and accurate substance identification. CAS roles have been modified effective December 16, 2001. Please check your SDI profiles to see if they need to be revised. For information on CAS roles, enter HELP ROLES at an arrow prompt or use the CAS Roles thesaurus (/RL field) in this file. => tetrasulfonate 424 TETRASULFONATE **33 TETRASULFONATES** L1440 TETRASULFONATE (TETRASULFONATE OR TETRASULFONATES) => disodium 26655 DISODIUM 2 DISODIUMS 26657 DISODIUM L2 (DISODIUM OR DISODIUMS) => 11 and 12 LЗ 6 L1 AND L2 => sodium chloride 747538 SODIUM 31 SODIUMS 747551 SODIUM (SODIUM OR SODIUMS) 843535 CHLORIDE 127339 CHLORIDES 906955 CHLORIDE (CHLORIDE OR CHLORIDES) T.4 83230 SODIUM CHLORIDE (SODIUM(W)CHLORIDE) => 12 and 14 1783 L2 AND L4 L5 => 13 and 14 Lб 0 L3 AND L4 => d 13 1-6 ti T.3 ANSWER 1 OF 6 CAPLUS COPYRIGHT 2002 ACS Sun protection beyond the sun protection factor - new efficient and ΤI photostable UV-A filters L3 ANSWER 2 OF 6 CAPLUS COPYRIGHT 2002 ACS ΤI Reactions of indigosulfonates with carbonyl compounds L3 ANSWER 3 OF 6 CAPLUS COPYRIGHT 2002 ACS





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of the conventional, unstable UV-A Filter BMDBM (Bu Methoxydibenzoylmethane) with the modern photostable UV-A Filter MBBT. The new photostable UV-A/ Broadband filters are discussed and the first com. applications of their use are also taken into consideration. RE.CNT 43 THERE ARE 43 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT => DPBT L78 DPBT => d 17 1-8 ti ANSWER 1 OF 8 CAPLUS COPYRIGHT 2002 ACS L7 ΤI Sun protection beyond the sun protection factor - new efficient and photostable UV-A filters ANSWER 2 OF 8 CAPLUS COPYRIGHT 2002 ACS L7 ΤI Mercury(II) and gold(III) derivatives of 2-phenyl pyridines and 2-phenyl-4-(methylcarboxylato)quinoline ANSWER 3 OF 8 CAPLUS COPYRIGHT 2002 ACS 1.7 Thermal behaviours of Co(II), Ni(II), Cu(II), and Pb(II) complexes of ΤI N,N-dipropyl-N'-benzoylthiourea L7ANSWER 4 OF 8 CAPLUS COPYRIGHT 2002 ACS ΤI Allergens in rubber products ANSWER 5 OF 8 CAPLUS COPYRIGHT 2002 ACS T.7 ΤI A stable free radical as donor: a layer-structure organic pressure sensor T.7 ANSWER 6 OF 8 CAPLUS COPYRIGHT 2002 ACS ΤI A Pressure Sensitive Two-Dimensional Tetracyanoquinodimethane (TCNQ) Salt of a Stable Free Radical ANSWER 7 OF 8 CAPLUS COPYRIGHT 2002 ACS 1.7 Electrical conductivity of heterocyclic aromatic rigid-rod and ladder ΤI polymers: the effect of ion implantation L7 ANSWER 8 OF 8 CAPLUS COPYRIGHT 2002 ACS ΤI Developing a major pathway for peptide condensation: application of DPBT method in the synthesis of .alpha.-hANP => logoff hold COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION FULL ESTIMATED COST 17.56 17.77 DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) SINCE FILE TOTAL ENTRY SESSION CA SUBSCRIBER PRICE -0.62 -0.62 SESSION WILL BE HELD FOR 60 MINUTES STN INTERNATIONAL SESSION SUSPENDED AT 07:04:10 ON 05 JUN 2002

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FILE 'CAPLUS' ENTERED AT 06:59:22 ON 05 JUN 2002 L1 440 TETRASULFONATE L2 26657 DISODIUM L3 6 L1 AND L2 L4 83230 SODIUM CHLORIDE L5 1783 L2 AND L4 L6 0 L3 AND L4 L7 8 DPBT
=> ph
1111461 PH 8081 PHS L8 1115150 PH (PH OR PHS)
=> 11 and 18 . L9 75 L1 AND L8
=> hcl 541805 HCL 37 HCLS L10 541819 HCL (HCL OR HCLS)
=> 19 and 110 L11 2 L9 AND L10
=> d 111 1-2 ti
L11 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2002 ACS TI Reactive phthalocyanine dyes
L11 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2002 ACS TI Dihydropyridine compounds. IV. N-Phenyl-o-dihydropyridine and N-p-methoxyphenyl-o-dihydropyridine

=> d 19 65-75 ti

- L9 ANSWER 65 OF 75 CAPLUS COPYRIGHT 2002 ACS
- TI The oxidation-reduction stoichiometry and potential of microsomal cytochrome
- L9 ANSWER 66 OF 75 CAPLUS COPYRIGHT 2002 ACS TI Interaction of proteins and ions in solution. II. Association energy of serum albumin with organic anions
 - L9 ANSWER 67 OF 75 CAPLUS COPYRIGHT 2002 ACS
 - TI Reduction-oxidation potential of the juice of etiolated Vicia faba, and the changes in potential in some substrates under the action of enzymes from these plants
 - L9 ANSWER 68 OF 75 CAPLUS COPYRIGHT 2002 ACS TI Dehydrogenases of molds. I. An oxidizing glucose dehydrogenase from Aspergillus niger
 - L9 ANSWER 69 OF 75 CAPLUS COPYRIGHT 2002 ACS
 - TI Further investigations on the blood-brain barrier. The significance of the
 - electrical charge and the .zeta.-potential in the problem of the blood-brain barrier and capillary permeability in general
 - L9 ANSWER 70 OF 75 CAPLUS COPYRIGHT 2002 ACS
 - TI Dihydropyridine compounds. IV. N-Phenyl-o-dihydropyridine and N-p-methoxyphenyl-o-dihydropyridine
 - L9 ANSWER 71 OF 75 CAPLUS COPYRIGHT 2002 ACS TI Oxidation-reduction potentials of derivatives of thioindigo. I. Thioindigo

tetrasulfonate

- L9 ANSWER 72 OF 75 CAPLUS COPYRIGHT 2002 ACS
- TI Applications of oxidation-reduction potential
- L9 ANSWER 73 OF 75 CAPLUS COPYRIGHT 2002 ACS TI The colorimetric determination of oxidation-reduction potential and its application to brewing practice
- L9 ANSWER 74 OF 75 CAPLUS COPYRIGHT 2002 ACS TI Oxidation-reduction potentials of ascorbic acid
- L9 ANSWER 75 OF 75 CAPLUS COPYRIGHT 2002 ACS TI Apparent potential of solutions of reducing sugars

=> d 19 71,72 ti fbib abs

Thioindigo

L9 ANSWER 71 OF 75 CAPLUS COPYRIGHT 2002 ACS TI Oxidation-reduction potentials of derivatives of thioindigo. I. Thioindigo tetrasulfonate AN 1937:3024 CAPLUS DN 31:3024 OREF 31:398b-c TI Oxidation-reduction potentials of derivatives of thioindigo. I.

tetrasulfonate Preisler, Paul W.; Hempelmann, Louis H. AU J. Am. Chem. Soc. (1936), 58, 2305-8 SO DTJournal Unavailable LA The tetra-K salt of thioindigo tetrasulfonate (I) was prepd. AΒ after direct sulfonation of thioindigo with fuming H2SO4. The oxidation-reduction potentials of the system I-leuco-I was detd. for the pH range from 0 to 11.5. The orange-red oxidant is converted into the yellow reductant by a reaction involving 2 reduction equivs. per mole. The consts. of the system established are E0 = 0.409 v. and the first pK of the reductant is 4.5. Formation of a deeper red semiquinone was noted in the **pH** region from 8.5 to 11.5. Except in acid soln. of **pH** less than 2.5, the material is not particularly suitable as an oxidation-reduction indicator. L9 ANSWER 72 OF 75 CAPLUS COPYRIGHT 2002 ACS Applications of oxidation-reduction potential ΤI 1936:60813 CAPLUS AN DN 30:60813 OREF 30:8040h-i,8041a Applications of oxidation-reduction potential ΤT AU Deribere, M. SO Tech. moderne (1936), 28, 574-6 DTJournal LA Unavailable AB Commercial applications of rH are given in brewing, baking, wineries, tanneries, paper mills, photography and chem. analysis. At pH 7.0 50% reduction was detd. with a number of indicators to find the rH range: neutral red 2-4, phenosafranine 5-7, K indigodisulfonate 8-10, trisulfonate 9.5-11.5, tetrasulfonate 11-13, methylene blue 13.5-15.5, thionine 15-17, toluidine blue, 1-naphthol-2-dichlorophenol-2sulfonate of Na, and the analogous dibromo compd. 16-18; and phenol blue 20-22. For a series of indophenols (I) the following values were reported: 1-naphthol-2-I-sulfonate of Na and m-toluidinediamino-I 16.5-18.5, guaiacol-I-2,6-dibromophenol 17.5-19.5, thymol-I 18-20, o-cresol-I-2,6-dichlorophenol 18-20, o-cresol-I 19-21, and 20-22 for phenol-I-2,6-dichloro- or dibromophenol, o-chlorophenol-I-2,6dichlorophenol, phenol-I, or o-bromo- or o-chlorophenol-I. => 14 and 19 L12 0 L4 AND L9 => potassium chloride 428942 POTASSIUM 16 POTASSIUMS 428946 POTASSIUM (POTASSIUM OR POTASSIUMS) 843535 CHLORIDE 127339 CHLORIDES 906955 CHLORIDE (CHLORIDE OR CHLORIDES) L13 41063 POTASSIUM CHLORIDE (POTASSIUM(W)CHLORIDE) => 113 and 19 L14 0 L13 AND L9

=> d his (FILE 'HOME' ENTERED AT 06:59:12 ON 05 JUN 2002) FILE 'CAPLUS' ENTERED AT 06:59:22 ON 05 JUN 2002 440 TETRASULFONATE L1L2 26657 DISODIUM L3 6 L1 AND L2L4 83230 SODIUM CHLORIDE L51783 L2 AND L4 Lб 0 L3 AND L4 L78 DPBT L8 1115150 PH L9 75 L1 AND L8 L10 541819 HCL 2 L9 AND L10 L11L12 0 L4 AND L9 L13 41063 POTASSIUM CHLORIDE L14 0 L13 AND L9 => pk 20251 PK 1437 PKS 21487 PK L15 (PK OR PKS) => 11 and 115 L16 1 L1 AND L15 => d l16 ti fbib abs L16 ANSWER 1 OF 1 CAPLUS COPYRIGHT 2002 ACS TΤ Oxidation-reduction potentials of derivatives of thioindigo. I. Thioindigo tetrasulfonate ΔM 1937:3024 CAPLUS DN 31:3024 OREF 31:398b-c Oxidation-reduction potentials of derivatives of thioindigo. I. тr Thioindigo tetrasulfonate AU Preisler, Paul W.; Hempelmann, Louis H. SO J. Am. Chem. Soc. (1936), 58, 2305-8 DT Journal Γ.A Unavailable AB The tetra-K salt of thioindigo **tetrasulfonate** (I) was prepd. after direct sulfonation of thioindigo with fuming H2SO4. The oxidation-reduction potentials of the system I-leuco-I was detd. for the pH range from 0 to 11.5. The orange-red oxidant is converted into the yellow reductant by a reaction involving 2 reduction equivs. per mole. The consts. of the system established are E0 = 0.409 v. and the first $\mathbf{p}\mathbf{K}$ of the reductant is 4.5. Formation of a deeper red semiquinone was noted in the pH region from 8.5 to 11.5. Except in acid soln. of pH less than 2.5, the material is not particularly suitable as an oxidation-reduction indicator.

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